

Aging of Polyurethane Foam Insulation in Simulated Refrigerator Panels — Two-Year
Results with Third-Generation Blowing Agents*

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AGING OF POLYURETHANE FOAM INSULATION IN SIMULATED REFRIGERATOR PANELS — TWO-YEAR RESULTS WITH THIRD-GENERATION BLOWING AGENTS

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cyclopentane. Tests with these third-generation blowing agents started in the fall of 1997 and are planned to continue
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ABSTRACT

Laboratory data are presented on the effect of constant-temperature aging on the apparent thermal conductivity of polyurethane foam insulation for refrigerators and freezers. The foam specimens were blown with HCFC-141b and with three of its potential replacements — HFC-134a, HFC-245fa, and cyclopentane. Specimens were aged at constant temperatures of 90°F, 40°F, and -10°F. Thermal conductivity measurements were made on two types of specimens: full-thickness simulated refrigerator panels containing foam enclosed between solid plastic sheets, and thin slices of core foam cut from similar panels. Results are presented for the first two years of a multi-year aging study. Preliminary comparisons of measured data with predictions of a mathematical aging model are presented.

INTRODUCTION

Polyurethane foam insulation currently used in refrigerators and freezers in the United States is blown primarily with HCFC-141b. Because of its potential for depleting the ozone layer, the Montreal Protocol mandates that domestic production of HCFC-141b cease by the end of 2002, requiring that a replacement blowing agent be used after that time. Since 1993, the Oak Ridge National Laboratory has been cooperating with the Appliance Research Consortium on studies of the aging characteristics of polyurethane insulation foamed with various blowing agents. The early studies were aimed at the replacement for CFC-11, and results of aging studies on simulated refrigerator panels blown with CFC-11, HCFC-141b, and a blend of HCFC-142b and HCFC-22 have been reported previously [1,2]. A long-term study is currently underway on aging of foam blown with HCFC-141b and several potential replacements — HFC-134a, HFC-245fa, and

for the next few years. Results on full-thickness specimens after aging at controlled temperatures for six months and for one year have been reported previously [3,4]. This paper presents results after two years of aging.

SPECIMENS

Two types of specimens are being studied. Specimens of one type were fabricated as panels that simulate the construction of a door or wall of a refrigerator. The panels are about two inches thick and have lateral dimensions of 24 inches by 24 inches. The faces of the panels are bounded by solid sheets. For the previous study of second-generation blowing agents, the solid sheets consisted of 24 gauge (0.024 in. thick) steel on one side and 0.12 in. thick acrylonitrile-butadiene-styrene (ABS) plastic on the other side [1,2]. For the present study of third-generation blowing agents, 0.040 in. thick plastic sheets were used on both faces, with separate sets of panels being made with ABS and high-impact-polystyrene (HIPS) plastic. The thinner plastic was considered to be more representative of current refrigerator production, and plastic was used on both sides to accelerate the aging experiments by allowing gases to permeate through both sides of the panels. It was felt that the effect of a steel sheet on one side could be simulated using models that are being developed. The edges of the panels were sealed with aluminum foil tape to simulate the configuration in a refrigerator where there are no comparable cut exposed foam edges.

The panels were foamed with HCFC-141b (to provide the base case), HFC-245fa, HFC-134a, and cyclopentane. At the present time, the latter two blowing agents are the only commercially available non-ozone-depleting replacements for HCFC-141b, but commitments have been made to

construct a plant to produce HFC-245fa. Commercial availability of HFC-245fa is expected in mid-2002. The panels were fabricated by four foam suppliers (denoted here as suppliers A, B, C, and D), with each supplier providing panels with each of the four blowing agents.

To provide a characterization of the foam itself, specimens were also made that consisted of core foam cut into 12 in. squares and sliced into thicknesses of about 0.4 and 1.5 inches. A stack of four of the 0.4-in.-thick slices made up one test specimen.

It should be noted that the foams made with the alternative blowing agents may not yet be optimized for thermal performance. Visual observations of the sliced specimens showed a very uniform fine cellular structure with HCFC-141b but the presence of many larger bubbles interspersed within the fine cellular structure for the other blowing agents.

EXPERIMENTAL PROCEDURES

Thermal resistance measurements were made using heat-flow-meter-apparatuses (HFMA) that conform to ASTM C 518 [5]. Intervening layers of foam rubber were placed between the panel specimens and the hot and cold plates of the apparatus to eliminate any undesirable air gaps between the specimens and the plates and also to protect the plates from the rigid test panels. Thermocouples were taped directly to the faces of the panels so that the temperature differences across the test panels were measured directly. Since the measurements gave the overall thermal resistance of the center of the test panel, a small correction (< 1%) was made for the thermal resistance of the plastic sheets to obtain the apparent thermal conductivity of the foam insulation. Tests on the core foam specimens gave the apparent thermal conductivity directly.

The specimens were stored in closed, constant temperature, atmospheric pressure aging chambers between HFMA tests. Aging temperatures of 90°F, 40°F, and -10°F were used in order to span most of the range of conditions to which the foam would be exposed in a refrigerator application. Thermal tests were performed at 45°F and 75°F mean temperatures, using a temperature difference of 40°F. Thermal measurements were made on the full-thickness panels soon after their fabrication and then at six-month intervals. Since the slices of core foam aged much more rapidly than the full-thickness panels, more frequent thermal measurements were required. This limited the number of core-foam specimens that could be studied, and core-foam measurements were concentrated on specimens from Supplier

A.

RESULTS AND DISCUSSION

Core-Foam Specimens

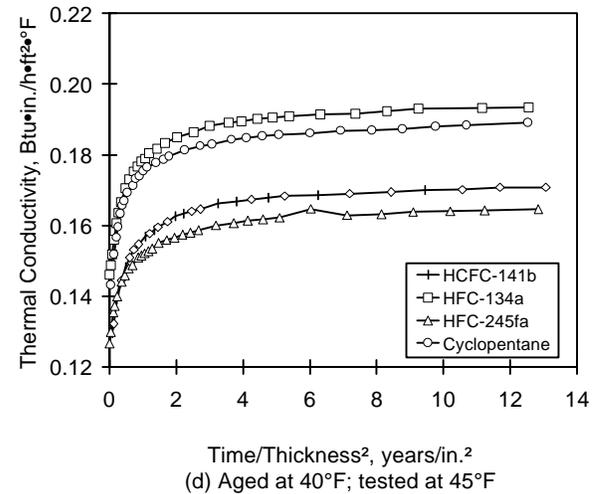
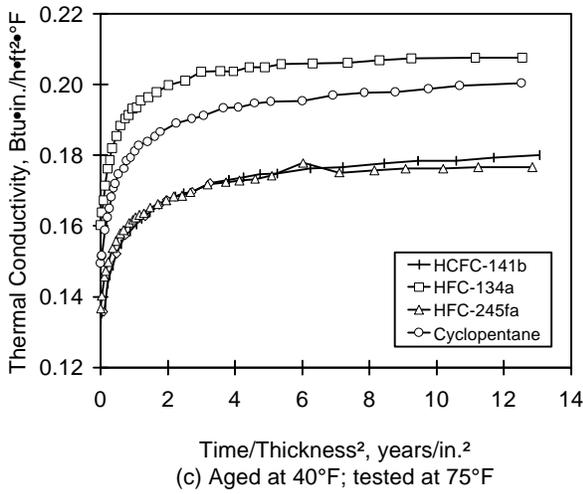
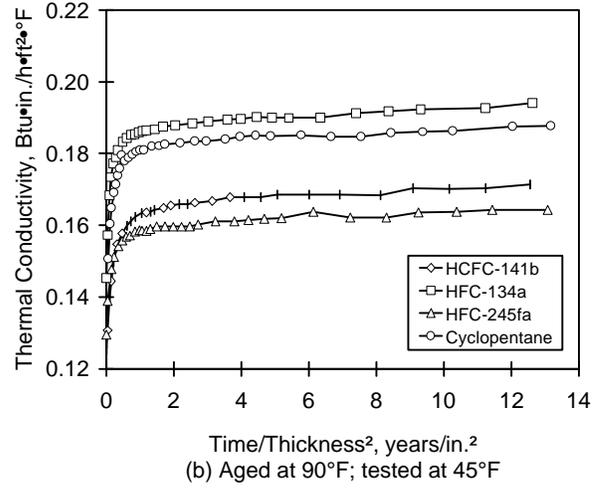
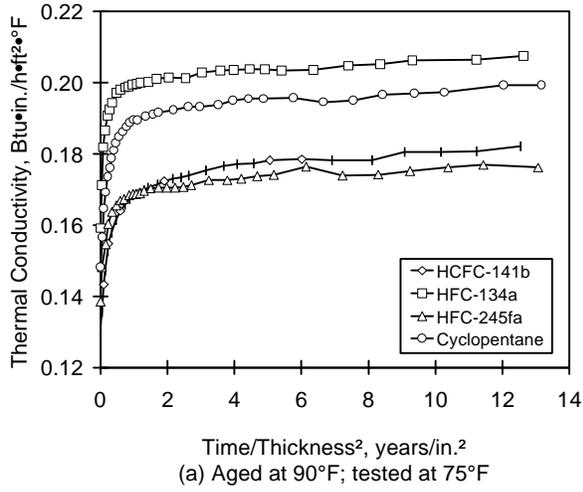
The thermal conductivity of freshly sliced foam at 75°F is shown in Table 1 for the two slice thicknesses. This shows that the lowest thermal conductivity was found with HCFC-141b, followed by HFC-245fa (4% higher), cyclopentane (13% higher), and HFC-134a (21% higher). This relative ranking is the same as has been observed by Haworth [6]. Table 1 also shows that the thermal conductivity of thinner slices was consistently 3 to 4% higher than that of the thicker slices. This is at least partly due to the larger amount of damaged surface layers with the thinner slices where air immediately displaces the blowing agent in the cut cells. While the data for the thicker slices are more representative of that in the full-thickness panels, the thinner specimens age much more rapidly and are useful for comparing potential differences in long-term performance with the different blowing agents.

Figure 1 compares the aging characteristics of the thinner (0.4 in.-thick) slices with the four blowing agents for three aging temperatures and for the two test temperatures. The data were taken over a period of about two years. Since numerous previous studies have shown that the aging of unfaced foam scales with the square of the thickness (see [3] and ASTM C 1303 [7]), the data are plotted as thermal conductivity versus the aging time divided by the square of the slice thickness. With this scaling, data obtained over a time period of two years on slices 0.4 in. thick are good predictors of the variation of the thermal conductivity of unfaced 2 in.-thick specimens over a period of 50 years, well beyond the nominal refrigerator lifetime of 20 years.

The curves in Figure 1 show several trends. For all conditions, the curves show an initial rapid increase in thermal conductivity which is attributed to diffusion of air components into the cells of the foam, followed by a more gradual increase which is attributed to diffusion of the blowing agent out of the cells. Except for crossover of some of the curves for HCFC-141b and HFC-245fa, the relative ranking of blowing agents given in Table 1 is preserved throughout the aging process. The time required for a given change in thermal conductivity with aging at 40°F is roughly twice as long as for aging at 90°F, while the time required at -10°F is about 10 times as long as at 90°F. The curves for tests at 45°F and 75°F mean temperatures show the same

Table 1. Thermal Conductivity of Freshly-Sliced Core-Foam Specimens at 75°F

Slice Thickness, in.	HCFC-141b	HFC-134a	HFC-245fa	Cyclopentane



0.4	0.132	0.160	0.138	0.150
1.5	0.128	0.155	0.132	0.145

Thermal conductivity units are $\text{Btu}\cdot\text{in.}/\text{h}\cdot\text{ft}^2\cdot^\circ\text{F}$. Each value represents an average over three specimens. Specimens are from Supplier A.

Figure 1. Aging of thin (0.4 in.-thick) core-foam specimens blown with third-generation blowing agents by Supplier A. Captions show aging and testing temperatures.

general variations, but the values measured at 45°F are always lower than those measured at 75°F.

Data read from the curves in Figure 1 and companion data for the thicker (1.5 in.-thick) slices are compared with data obtained on full-thickness panels in the next section. In general, the data obtained on core-foam specimens may be used to estimate upper bounds for aging of the foam in the full-thickness panels.

Full-Thickness Panels

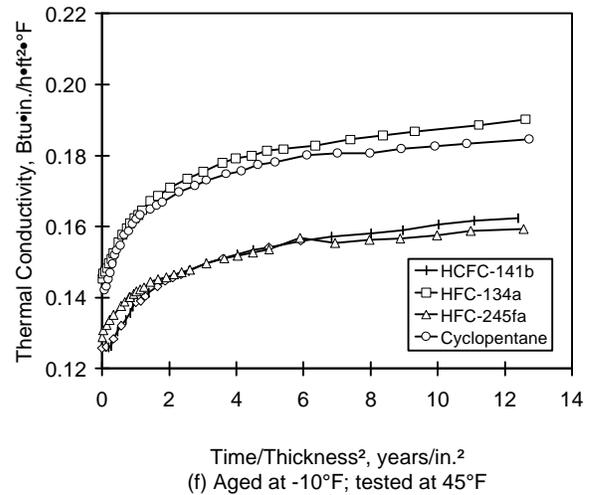
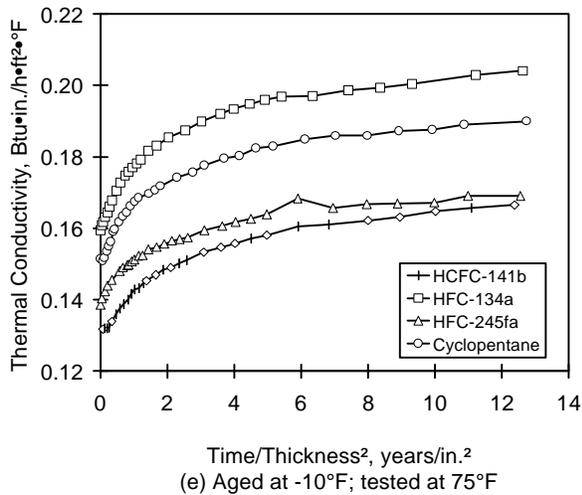
Thermal measurements on 96 full-thickness panels have been performed before and after aging under controlled temperatures for two years. Data on the panels are reported as normalized values because the directly measured values are biased due to the construction of the panels. From tests on standard reference materials, we have found that heat flows due to the aluminum foil tape around the panel edges result in measured center-of-panel thermal conductivities that are too low by a few percent. The data were normalized by averaging the pre-aging thermal conductivities measured at 75°F on the 24 panels that were blown with HCFC-141b, and then dividing the individual measured thermal conductivity values by this average. This normalization procedure was considered to be justified since the bias caused by the aluminum foil tape should be nearly the same for all panels tested.

Normalized results before aging and after aging for one and two years are given in Table 2. The results in the table give an indication of the variation in thermal conductivity among specimens from a particular supplier and among the

the variation is 2% or less. Similar levels of variation are seen for the other blowing agents.

Averaging the pre-aging data on the six panels for each supplier and each blowing agent gives the comparisons shown in Figure 2. On average, the 75°F thermal conductivity of foam blown with HFC-134a was about 18% higher than for HCFC-141b, while the conductivities for HFC-245fa and cyclopentane were about 7% and 15% higher. For Supplier A only, the average conductivities for HFC-134a, HFC-245fa, and cyclopentane were 20%, 5%, and 15% higher, respectively, than for HCFC-141b. These relative differences are in good agreement with the values given in Table 1 for the core foam, also from Supplier A. At the 45°F test temperature, the average conductivities for HFC-134a, HFC-245fa, and cyclopentane were 16%, 4%, and 16% higher than for HCFC-141b.

Figure 3 shows the variation with time for panels from Supplier A as they were aged at 90°F and 40°F and tested at 75°F. Data for -10°F aging are not shown because the changes were very small. Figure 3 shows that the panels with ABS sheets aged less rapidly than did the panels with HIPS sheets, and that aging at 40°F was slower than at 90°F. Figure 3 also shows that the relative ranking of the four blowing agents is maintained during aging over the two year period, except for the HIPS panels aged at 90°F, where the curves for HCFC-141b and HFC-245fa crossed and also where the curves for HFC-134a and cyclopentane came together. The shape of the curves in Figure 3b is similar to those of Figure 1, where the increase in conductivity is more rapid at shorter times. The other curves are more or less linear, but would be expected to develop the same shape as



four different suppliers. For example, the panels blown with HCFC-141b have pre-aging normalized conductivities at 75°F that range from 0.95 to 1.04, but for a given supplier

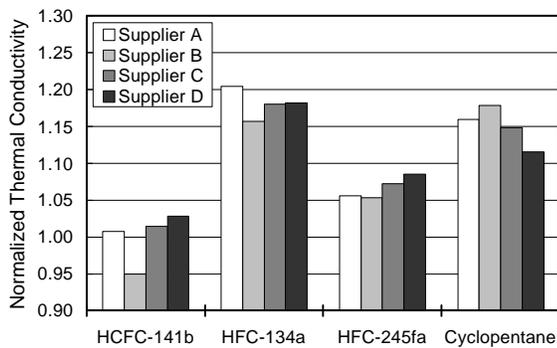
in Figure 3b as aging progresses.

Figure 4 shows percentage increases in thermal conductivity after two years of aging. The values were

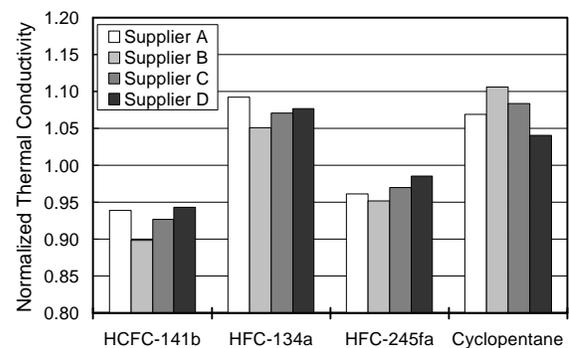
obtained by averaging the percentage increases for the four suppliers (except for HFC-245fa with HIPS at -10°F, where the anomalously high values for Supplier B were not

included in the average). Also shown for comparison are the percentage increases for 2 in.-thick core foam. The values at

Figure 2. Average normalized thermal conductivity for full-thickness panels before aging. Data are normalized with respect to average for HCFC-141b tests at 75°F before aging.



(a) Tested at 75°F



(b) Tested at 45°F

Table 2. Results of thermal conductivity tests on full thickness panels before and after two years of aging at controlled temperature.

Values are normalized to the average of the pre-aging results for HCFC-141b at 75°F test temperature.

Blowing		Aging	Tested at 75°F												Tested at 45°F											
Agent	Liner	Temp., °F	Supplier A			Supplier B			Supplier C			Supplier D			Supplier A			Supplier B			Supplier C			Supplier D		
			0 Yr.	1 Yr.	2 Yr.	0 Yr.	1 Yr.	2 Yr.	0 Yr.	1 Yr.	2 Yr.	0 Yr.	1 Yr.	2 Yr.	0 Yr.	1 Yr.	2 Yr.	0 Yr.	1 Yr.	2 Yr.	0 Yr.	1 Yr.	2 Yr.	0 Yr.	1 Yr.	2 Yr.
HCFC-141b	ABS	90	1.01	1.06	1.13	0.95	1.00	1.06	1.03	1.09	1.17	1.04	1.09	1.16	0.94	0.97	1.03	0.90	0.92	0.99	0.93	0.98	1.05	0.95	0.99	1.06
HCFC-141b	ABS	40	1.01	1.04	1.06	0.95	1.00	1.04	1.01	1.07	1.12	1.03	1.08	1.11	0.94	0.93	0.95	0.91	0.90	0.94	0.92	0.95	0.99	0.95	0.96	0.99
HCFC-141b	ABS	-10	1.01	1.02	1.03	0.95	0.96	0.97	1.02	1.06	1.06	1.04	1.05	1.06	0.94	0.93	0.93	0.89	0.88	0.88	0.93	0.93	0.93	0.95	0.94	0.94
HCFC-141b	HIPS	90	1.01	1.13	1.21	0.95	1.06	1.13	1.01	1.18	1.25	1.02	1.19	1.26	0.94	1.04	1.11	0.90	0.99	1.06	0.93	1.07	1.13	0.94	1.10	1.16
HCFC-141b	HIPS	40	1.00	1.06	1.11	0.95	1.01	1.06	1.01	1.13	1.18	1.03	1.11	1.20	0.94	0.97	1.01	0.90	0.92	0.97	0.93	1.02	1.06	0.94	1.02	1.09
HCFC-141b	HIPS	-10	1.01	1.03	1.03	0.95	0.95	0.97	1.01	1.07	1.08	1.02	1.05	1.07	0.94	0.94	0.95	0.90	0.88	0.88	0.93	0.96	0.96	0.94	0.95	0.96
HFC-134a	ABS	90	1.21	1.26	1.33	1.17	1.23	1.28	1.17	1.25	1.31	1.18	1.25	1.31	1.09	1.13	1.20	1.07	1.11	1.17	1.06	1.12	1.18	1.08	1.13	1.19
HFC-134a	ABS	40	1.20	1.23	1.28	1.14	1.19	1.21	1.17	1.24	1.27	1.19	1.23	1.26	1.09	1.10	1.13	1.04	1.05	1.07	1.07	1.10	1.12	1.08	1.10	1.12
HFC-134a	ABS	-10	1.21	1.22	1.24	1.16	1.18	1.18	1.18	1.22	1.22	1.18	1.22	1.23	1.09	1.09	1.10	1.05	1.06	1.06	1.07	1.09	1.07	1.08	1.10	1.09
HFC-134a	HIPS	90	1.20	1.33	1.40	1.15	1.28	1.33	1.19	1.33	1.39	1.17	1.30	1.37	1.09	1.21	1.27	1.04	1.16	1.22	1.08	1.20	1.27	1.07	1.19	1.24
HFC-134a	HIPS	40	1.21	1.27	1.32	1.14	1.20	1.25	1.18	1.27	1.30	1.18	1.24	1.28	1.09	1.13	1.18	1.04	1.07	1.13	1.08	1.13	1.16	1.08	1.11	1.14
HFC-134a	HIPS	-10	1.21	1.23	1.26	1.17	1.21	1.21	1.19	1.24	1.25	1.19	1.22	1.23	1.09	1.10	1.12	1.07	1.08	1.08	1.07	1.10	1.11	1.08	1.09	1.09
HFC-245fa	ABS	90	1.06	1.10	1.14	1.04	1.07	1.12	1.07	1.12	1.18	1.09	1.13	1.18	0.97	0.98	1.04	0.95	0.96	1.01	0.97	1.00	1.06	0.99	1.02	1.06
HFC-245fa	ABS	40	1.05	1.08	1.11	1.07	1.11	1.18	1.07	1.11	1.13	1.09	1.13	1.15	0.96	0.96	0.99	0.96	0.97	1.07	0.97	0.97	0.99	0.99	1.00	1.01
HFC-245fa	ABS	-10	1.06	1.06	1.08	1.04	1.06	1.12	1.07	1.10	1.07	1.09	1.12	1.12	0.96	0.96	0.97	0.94	0.94	0.99	0.96	0.97	0.94	0.99	1.00	0.99
HFC-245fa	HIPS	90	1.05	1.15	1.18	1.05	1.13	1.18	1.08	1.18	1.24	1.08	1.18	1.23	0.96	1.04	1.08	0.94	1.01	1.07	0.97	1.07	1.12	0.98	1.07	1.12
HFC-245fa	HIPS	40	1.06	1.11	1.15	1.05	1.09	1.14	1.08	1.14	1.17	1.08	1.15	1.18	0.97	0.99	1.03	0.95	0.96	1.01	0.97	1.01	1.03	0.99	1.03	1.05
HFC-245fa	HIPS	-10	1.05	1.08	1.08	1.08	1.31	1.40	1.07	1.11	1.12	1.09	1.11	1.12	0.96	0.97	0.97	0.98	1.20	1.28	0.97	0.99	1.00	0.98	1.00	1.00
Cyclopentane	ABS	90	1.16	1.22	1.30	1.18	1.22	1.29	1.14	1.22	1.28	1.11	1.18	1.27	1.07	1.13	1.22	1.10	1.16	1.22	1.07	1.17	1.22	1.04	1.11	1.19
Cyclopentane	ABS	40	1.16	1.19	1.24	1.18	1.21	1.23	1.14	1.21	1.23	1.12	1.16	1.19	1.06	1.07	1.11	1.10	1.12	1.14	1.07	1.11	1.14	1.04	1.05	1.08
Cyclopentane	ABS	-10	1.16	1.17	1.18	1.18	1.20	1.21	1.14	1.20	1.19	1.12	1.14	1.14	1.06	1.05	1.06	1.10	1.10	1.11	1.07	1.10	1.09	1.05	1.04	1.04
Cyclopentane	HIPS	90	1.17	1.33	1.39	1.18	1.30	1.36	1.16	1.32	1.37	1.11	1.28	1.35	1.09	1.25	1.31	1.12	1.25	1.30	1.10	1.27	1.31	1.02	1.22	1.28
Cyclopentane	HIPS	40	1.15	1.24	1.31	1.18	1.24	1.27	1.15	1.26	1.31	1.12	1.23	1.27	1.06	1.13	1.20	1.11	1.16	1.19	1.09	1.19	1.23	1.05	1.14	1.19
Cyclopentane	HIPS	-10	1.16	1.19	1.21	1.17	1.19	1.20	1.15	1.21	1.23	1.12	1.16	1.18	1.07	1.07	1.11	1.11	1.11	1.11	1.09	1.14	1.15	1.05	1.07	1.09

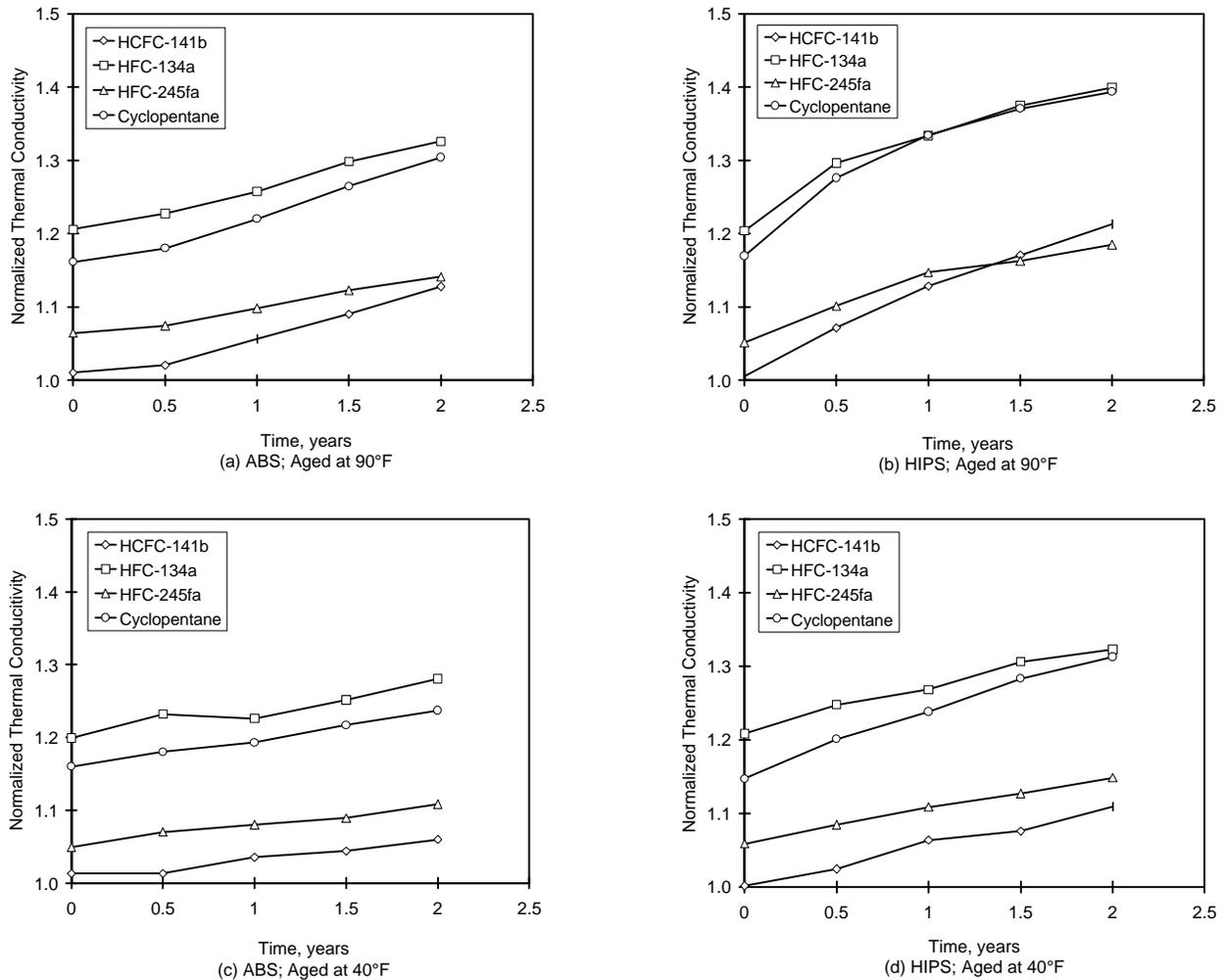


Figure 3. Aging of foam in full-thickness panels blown with third-generation blowing agents by Supplier A. Tested at 75°F.

two years and 20 years of aging were obtained by scaling the results for 1.5 in.-thick and 0.4 in.-thick core-foam specimens, respectively. The results show that the plastic sheets significantly reduce the rate of aging, with ABS being more effective than HIPS. The difference between ABS and HIPS may be explained by the relative gas permeances of the two plastics, as will be shown in the section on modeling. Figure 4 also shows that decreasing temperature produces large decreases in the rate of aging, with the changes at 40°F being about one-half as large as at 90°F and with the changes at -10°F being very small. This is in agreement with the effect of temperature observed on core foam, and also in agreement with the effect of temperature on the permeance of the plastic sheets and for coefficients of diffusion of gases through the foam, as will be shown in the section on modeling.

The data presented for the test panels should not be interpreted directly as quantitative indications of the changes that would be expected in the walls and doors of refrigerators. This is because only one surface in a

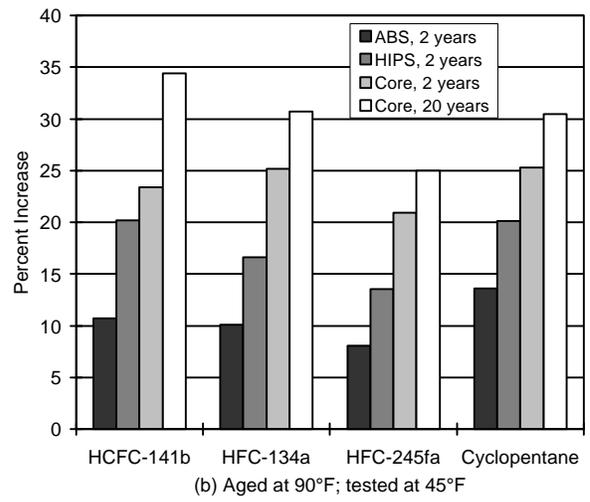
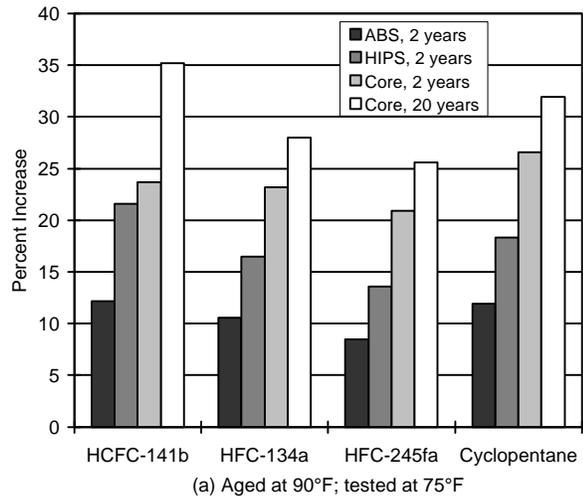
refrigerator will have a plastic sheet, while the other surface will have an impermeable steel sheet. This effect is estimated in the section on modeling.

COMPARISON OF DATA WITH MODEL RESULTS

Since aging data on the full-thickness panels are being collected on a real-time basis, there is a need for a model that can predict long-term aging behavior. A one-dimensional model for isothermal aging of foam panels with semi-permeable surface sheets has been developed, and is being validated against the measured aging data. Preliminary results from the model are presented here.

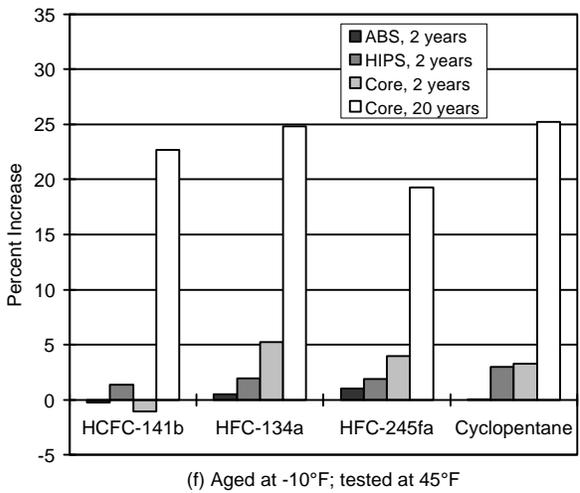
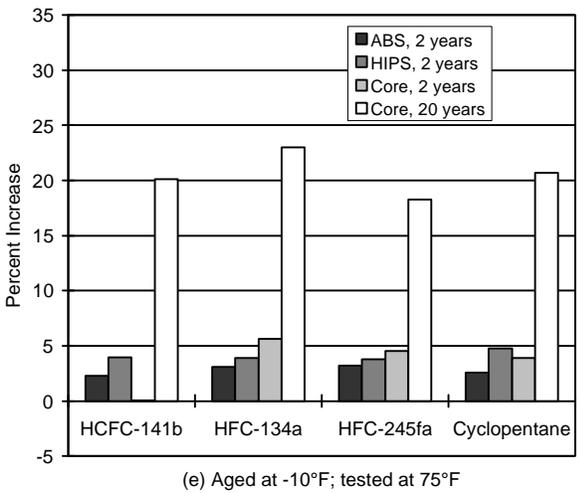
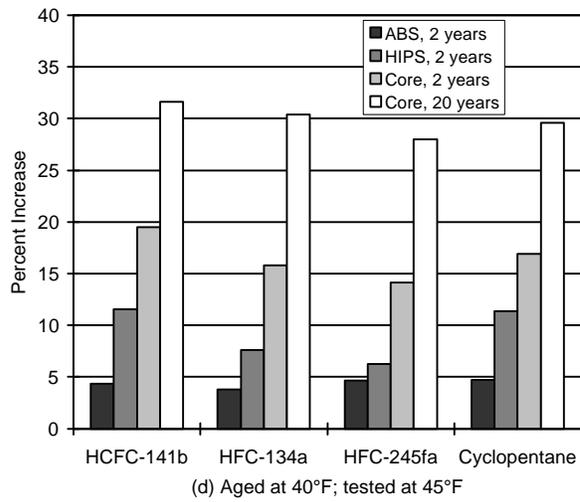
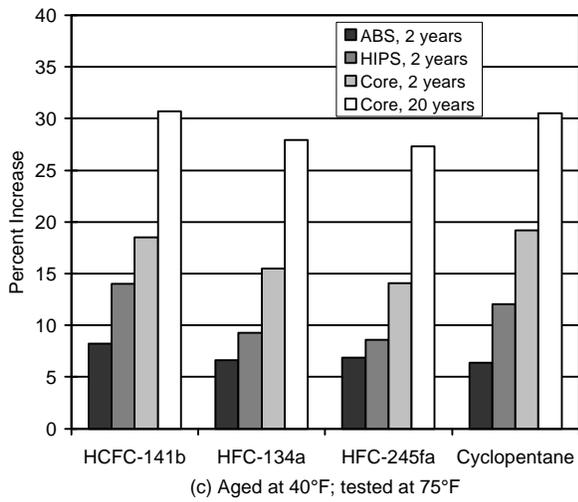
Description of Model

The model is similar to those described in the literature for diffusion of gases through foam [e.g. 8]. However, we have added the effects of partial condensation of blowing agent within the foam cells. The transient distribution of a gas



during aging is given by Fick's second law of diffusion:

Figure 4. Percentage increase in thermal conductivity of foam in full-thickness panels after two years of aging. The legend shows the surface material: ABS or HIPS plastic sheets. Values for panels are averages over four foam suppliers. Values for core foam are scaled from aging curves for slices of foam from Supplier A.



$$\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2} \quad (1)$$

where c is the concentration of gas in moles per unit volume, D is the effective diffusion coefficient which is taken to be a constant, t is time, and x is distance. The gas concentration (c) and pressure (P) are related through the ideal gas law, $c = P/RT$, where R is the gas constant, and T is the absolute temperature. The foam is produced at relatively high temperatures, but the panels are being aged at lower temperatures, where a fraction of the blowing agent can be in a condensed state. In this case, it is assumed that the liquid is immobile and the only mechanism for transport of the blowing agent is by vapor diffusion. The vapor pressure curve provides an upper limit on the gas pressure, and a mass balance is added to track the distribution of liquid and vapor phases of the blowing agent.

The initial conditions for Equation 1 are that the concentrations of oxygen and nitrogen are zero, while the concentrations of carbon dioxide and blowing agent are uniform, with values that are provided by the foam supplier. Boundary conditions for Equation 1 are given by application of Fick's first law of diffusion:

$$-D \frac{\partial c}{\partial x} = K (c_0 - c) \quad (2)$$

where K is the permeance of a surface sheet ($K = \infty$ for bare cut foam, 0 for solid metal surface, and intermediate values for plastic sheets on the surface), c_0 is the concentration of gas in the space surrounding the panel, and c and the derivative are evaluated at the interface between the foam and the surface sheet. The model has been formulated to allow for different permeances on each side of the panel. The surroundings were taken to consist of nitrogen and oxygen at concentrations that correspond to 0.21 atm for oxygen and 0.79 atm for nitrogen.

Equations 1 and 2 are used to calculate concentrations through the thickness of the specimen during aging. Gas pressure profiles under the thermal test conditions are then calculated from the concentration profiles using the ideal gas law and the vapor pressure curve for the blowing agent. Next, the Lindsay-Bromley formulation of the Wassiljewa equation is used to calculate the thermal conductivity of the gas mixture as a function of position through the specimen [9]. The Wassiljewa equation is

$$k_{gas} = \frac{\sum_{i=1}^N y_i k_{gi}}{\sum_{j=1}^N y_j A_{ij}} \quad (3)$$

where k_{gas} is the thermal conductivity of the gas mixture, k_{gi}

and y_i are the thermal conductivity and mole fraction of the i^{th} gas, and A_{ij} is given by the Lindsay-Bromley equation,

$$A_{ij} = \frac{1}{4} \left[1 + \left[\frac{\mu_i}{\mu_j} \left(\frac{M_j}{M_i} \right)^{0.75} \frac{\left(1 + \frac{S_{ij}}{T} \right)}{\left(1 + \frac{S_j}{T} \right)} \right]^{1/2} \right]^2 \frac{\left(1 + \frac{S_{ij}}{T} \right)}{\left(1 + \frac{S_i}{T} \right)} \quad (4)$$

where μ_i , M_i , and S_i are the dynamic viscosity, molecular weight, and Sutherland constant for the i^{th} gas, and T is the absolute temperature. Also, $S_i = 1.5 T_{bi}$, where T_{bi} is the normal boiling point of the i^{th} gas, and $S_{ij} = (S_i S_j)^{1/2}$ for non-polar gases. Gas properties needed for specimens blown with HCFC-141b are given in Table 3.

The total apparent thermal conductivity, k , at a plane within the specimen is then calculated from the equation given by Glicksman [15] as

$$k = k_{sol} + k_{rad} + \delta k_{gas} = k_0 + \delta k_{gas} \quad (5)$$

where k_{sol} represents conduction through the solid polymer cell walls and struts, k_{rad} represents radiation through the foam, δ is the void fraction (taken to be about 0.97), and k_{gas} is the thermal conductivity of the gas mixture. Glicksman gives formulae for estimating the values of k_{sol} and k_{rad} ; however, for the purposes of this study, their sum is estimated from the measured thermal data. The average thermal conductivity of the specimen is then calculated from an integral over the thickness (L) of the specimen,

$$k_{avg} = \left[\frac{1}{L} \int_0^L \frac{1}{k(x)} dx \right]^{-1} \quad (6)$$

Material Properties for Model

In addition to properties of the diffusing gases, the model requires values of the effective diffusion coefficients for the foam and gas permeances for the surface sheets. Permeances of CO_2 , O_2 , and N_2 through specimens of the 0.040 in.-thick ABS and HIPS plastic sheets were measured by a commercial laboratory [16], and the results are given in Table 4. These results show that the permeances for HIPS are four to six times larger than for ABS, and that the permeances at 40°F were 0.4 to 0.7 of the 90°F values. Diffusion coefficients of gases through specimens of the foam insulation are in the process of being measured using a gravimetric method that was developed by Booth [17]. Data that have been obtained to date are given in Table 5. The large differences among the gases should be noted with both the permeances and diffusion coefficients being ordered as $\text{CO}_2 > \text{O}_2 > \text{N}_2$. Values of these two parameters at

temperatures other than those at which values were measured were estimated from the Arrhenius equation, as, for example,

Gas	Thermal Conductivity, Btu•in./h•ft ² •°F	Viscosity, centipoise*	Molecular Weight	Normal Boiling Point, K
Nitrogen	0.178 [10]	0.0174	28.01 [14]	77.4 [14]
Oxygen	0.179 [10]	0.0200	32.00 [14]	90.2 [14]
Carbon dioxide	0.115 [11]	0.0148	44.01 [14]	194.7** [14]
HCFC-141b	0.067 [12]	0.0102	116.95 [14]	305.2 [14]

* Estimated by Corresponding-States method [13].

** Normal sublimation point.

Gas	HIPS		ABS	
	40°F	90°F	40°F	90°F
CO ₂	55.6	82.9	8.78	13.6
O ₂	8.47	16.2	1.92	4.32
N ₂	1.64	3.58	<0.32	0.78

Units are cc(STP)/100 in.²•atm•day

Foam Blowing Agent	Diffusing Gas	77°F (25°C)	104°F (40°C)
Cyclopentane	CO ₂	2.69 × 10 ⁻⁶	4.52 × 10 ⁻⁶
HCFC-141b	O ₂	3.30 × 10 ⁻⁷	5.03 × 10 ⁻⁷
HCFC-141b	N ₂	1.77 × 10 ⁻⁸	3.14 × 10 ⁻⁸

Units are cm²/s

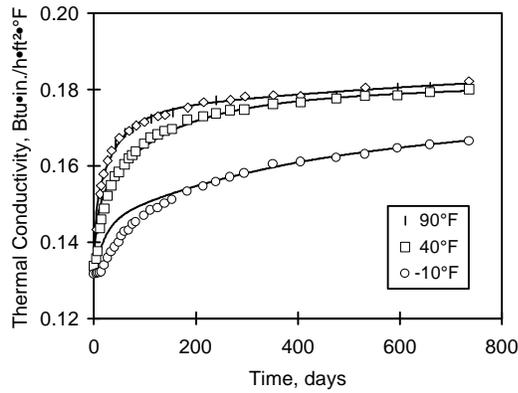
$$D = D_0 e^{-\frac{A}{RT}} \quad (7)$$

Values of D₀ and A were evaluated from the data at two temperatures.

Comparison of Model with Data

Figure 5 shows a comparison of model predictions with measured thermal data for the thinner slices of core foam blown with HCFC-141b. In addition to the parameters given

above, the model used only two other adjustable parameters. These are the coefficient of diffusion for HCFC-141b, and the sum of the radiative and solid contributions to the thermal conductivity (k₀ in Eq. 5). For aging of the thinner slices at 90°F, the model predicted that diffusion of all gases except HCFC-141b had been completed by about 300 days. After this, the slight increase in thermal conductivity is due to loss of the blowing agent. By a trial and error process, a diffusion coefficient of about 2.5 × 10⁻¹⁰ cm²/s was found to provide a good match for the data. For other temperatures, the ratio of diffusion coefficients for the blowing agent and nitrogen was assumed to be the same as at 90°F. This assumption is not critical, since the importance of diffusion



of the blowing agent decreases markedly with decreasing aging temperature, due partly to a decrease in the diffusion coefficient, but also due to a large decrease in blowing agent gas pressure because of condensation. At -10°F , the shape of the curve after about 200 days is mostly due to diffusion of nitrogen. Variation of k_0 shifts the curves up and down, but does not change their shape. Values of 0.065, 0.066, and 0.061 $\text{Btu}\cdot\text{in.}/\text{h}\cdot\text{ft}^2\cdot^{\circ}\text{F}$ were used for the comparisons shown in Figure 5. This Figure 5. Comparison of model predictions with measured data for thin (0.4 in. thick) specimens of foam blown with HCFC-141b from Supplier A. Symbols are measured data; lines are model predictions.

comparison shows that with a minimum of adjustable parameters, the model predicts the aging of core foam very well. This gives confidence that the independently-measured diffusion coefficients appropriately describe the aging mechanisms in the foam.

Figure 6 compares model predictions with measured data for the percentage increases in thermal conductivity for the HCFC-141b-blown foam contained in the full-thickness panels. The model simulations included a period of time between the production of the foam and the start of controlled-temperature aging, during which the panels were stored near room temperature. Figure 6a shows that the model predicts the aging of the panels with ABS surface sheets very well. Aging of the panels is controlled mostly by diffusion of CO_2 , O_2 , and N_2 . The curvature at short times is due to diffusion of CO_2 out of the foam, which causes an initial decrease of thermal conductivity by a few percent. The increases are mainly due to diffusion of oxygen and then nitrogen into the foam, with very little change in the concentration of blowing agent over this time period.

Figure 6b shows a similar comparison for the panels with HIPS surfaces. The agreement between the model and data is very good at 40°F and -10°F . At 90°F , there is good qualitative agreement, but the model underpredicts the aging by several percentage points. The agreement could be improved by allowing the gas permeances to be adjustable parameters. Comparisons on more panels, and over a longer

time period are needed for a more thorough assessment of the adequacy of the model for predicting long-term aging of panels.

Physical arguments suggest that the time to produce a given change in thermal conductivity would be two to four times longer when one plastic sheet is replaced with a steel sheet. The factor of two would apply if the permeance of the plastic sheets were the dominating resistance to gas transport, while the factor of four would apply if diffusion through the foam were the dominating resistance. The model was used to estimate the effect of a steel sheet on one side. For HIPS on one side and steel on the other, the model estimated that a time period about three times longer would be needed to produce the same change as with HIPS on both sides for aging at 40°F or 90°F . At -10°F , the model predicted a factor of 2.5. Similar simulations for ABS gave a factor of 2.4 to 2.5 for all three temperatures. These results show that for HIPS, the gas transport is controlled approximately equally by permeance through the plastic sheet and by diffusion through the foam. For ABS, gas transport is controlled more by permeance through the plastic sheet.

SUMMARY AND CONCLUSIONS

Thermal conductivity measurements have been made over a two year period on cut slices of polyurethane foam insulation blown with HCFC-141b, HFC-134a, HFC-245fa, and cyclopentane. Initial results at 75°F mean temperature before aging showed that the thermal conductivities of foam

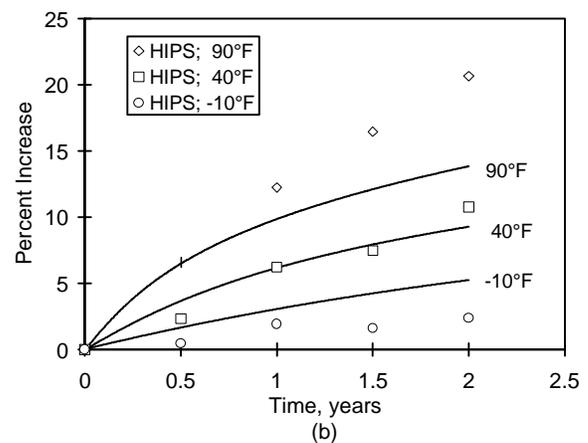


Figure 6. Comparison of model predictions with measured data for full-thickness panels containing foam blown with HCFC-141b from Supplier A. Symbols are measured data; lines are model predictions.

blown with HFC-134a, HFC-245fa, and cyclopentane were 21%, 4%, and 13%, respectively, higher than with HCFC-141b. The aging rate was very sensitive to aging temperature, with aging at 40°F being about one-half as fast as at 90°F, and with aging at -10°F being about one-tenth as fast. Except for crossover of the curves for HCFC-141b and HFC-245fa, the relative ranking of the blowing agents was preserved through the aging process.

Thermal conductivity measurements have been made on a set of full-thickness test panels containing polyurethane foam confined between solid sheets made of ABS and HIPS plastics. The panels simulated the walls and doors of a refrigerator or freezer except that the steel sheet normally on one side was replaced with a plastic sheet. Initial measurements at 75°F mean temperature before controlled aging showed that the conductivity of foam blown with HFC-134a, HFC-245fa, and cyclopentane averaged 18%, 7%, and 15%, respectively, higher than that with HCFC-141b.

Tests on the panels after two years of aging at controlled temperatures have been completed. For both ABS and HIPS plastics, the conductivity increases were less than those predicted for unenclosed full-thickness core-foam, showing that the plastic liners reduce the rate of aging. The panels with HIPS sheets showed average increases of 13 to 22% with aging at 90°F, 6 to 14% at 40°F, and less than 5% at -10°F. The panels with ABS sheets showed smaller increases of 8 to 14% at 90°F, 4 to 8% at 40°F, and less than 5% at -10°F. These differences in aging rates correlate with measurements of gas permeances of the plastic sheets.

Preliminary comparisons of a mathematical model with aging data show that the model is capable of predicting the aging behavior of thin slices of core foam very well. The model is also able to give good qualitative predictions of the aging of full-thickness panels with solid plastic surfaces. In most cases, very good quantitative agreement was obtained

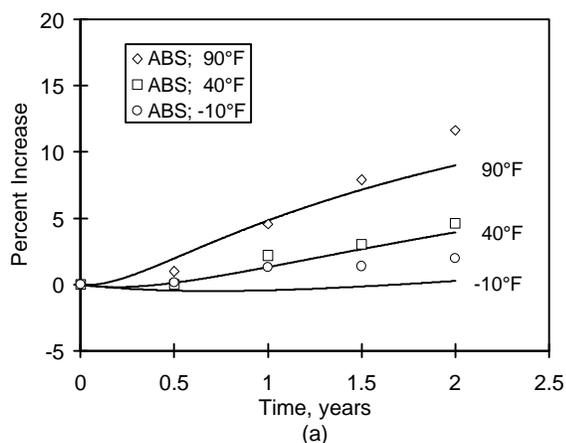
a more thorough assessment of the adequacy of the model for predicting long-term aging of panels.

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REFERENCES

1. Wilkes, K.E., D. W. Yarbrough, and F. J. Weaver. 1997. "Aging of Polyurethane Foam Insulation in Simulated Refrigerator Walls," *International Conference on Ozone Protection Technologies*, pp. 253-262.
2. Wilkes, K. E., W. A. Gabbard, and F. J. Weaver. 1998. "Aging of Polyurethane Insulation Foamed with Second- and Third-Generation Blowing Agents," *Proceedings of the 49th Annual International Appliance Technical Conference*, pp. 387-398.
3. Wilkes, K. E., W. A. Gabbard, and F. J. Weaver. 1998. "Aging of Polyurethane Foam Insulation in Simulated Refrigerator Panels - Initial Results with Third-generation Blowing Agents," *The Earth Technologies Forum*, pp. 280-289.
4. Wilkes, K. E., W. A. Gabbard, and F. J. Weaver. 1999. "Aging of Polyurethane Foam Insulation in Simulated Refrigerator Panels - One-Year Results with Third-generation Blowing Agents," *The Earth Technologies Forum*, pp. 303-313.
5. C 518, "Standard Test Method for Steady-State Heat Flux Measurements and Thermal Transmission Properties by Means of the Heat Flow Meter Apparatus," in *1999 Annual Book of ASTM Standards*. West Conshohocken, PA: American Society for Testing and Materials, Vol. 04.06, pp. 171-183.
6. Haworth, G. J. 1996. "Next Generation Insulation Foam Blowing Agents for Refrigerators/Freezers," *Proceedings*,



between the model and the full-thickness data. Comparisons on more panels, and over a longer time period are needed for

47th International Appliance Technical Conference, pp. 271-282.

7. C 1303, "Standard Test Method for Estimating the Long-Term Change in the Thermal Resistance of Unfaced Rigid Closed Cell Plastic Foam by Slicing and Scaling Under Controlled Laboratory Conditions," in *1999 Annual Book of ASTM Standards*. West Conshohocken, PA: American Society for Testing and Materials, Vol. 04.06, pp. 719-726.

8. Hoogendoorn, C. J. 1994. "Thermal Ageing," in *Low Density Cellular Plastics: Physical Basis of Behaviour*, N.

11. Eckert, E. R. G. and R. M. Drake, Jr. 1959. *Heat and Mass Transfer*. New York: McGraw-Hill, p. 506.

12. Shankland, I. R., D. J. Williams, M. C. Bogdan, and P. B. Logsdon. 1998. "Next Generation Blowing Agents for Thermal Insulating Foams," in *60 Years of Polyurethanes*, J. E. Kresta, E. W. Eldred, eds. Lancaster, PA: Technomic Publishing Co., Inc., p.86.

13. Reid, R. C. and T. K. Sherwood. 1966. *The Properties of Gases and Liquids*, 2nd Ed. New York: McGraw-Hill, p. 405.

14. *ASHRAE Handbook-Fundamentals*, 1997. Atlanta: American Society of Heating, Refrigerating, and Air-Conditioning Engineers, p. 18.3.

15. Glicksman, L. R. 1994. "Heat Transfer in Foams," in *Low Density Cellular Plastics: Physical Basis of Behaviour*, N. C. Hilyard and A. Cunningham, eds. London: Chapman & Hall, pp. 105-152.



16. Permeance measurements were made by Modern Controls, Inc., Minneapolis, Minnesota.

17. Booth, J. R., R. S. Graves, and D. W. Yarbrough. 1996. "Effective Diffusion Coefficients for CFC-11 by Gravimetric Depletion from Thin Slices of PIR Foams," in *Thermal Conductivity 23*, K. E. Wilkes, R. B. Dinwiddie, and R. S.

Graves, eds. Lancaster, PA: Technomic Publishing Co., Inc., pp. 352-337.

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C. Hilyard and A. Cunningham, eds. London: Chapman & Hall, pp. 153-186.

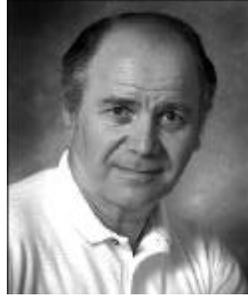
9. Reid, R. C. and T. K. Sherwood. 1966. *The Properties of Gases and Liquids*, 2nd Ed. New York: McGraw-Hill, pp. 483-484.

10. *ASHRAE Handbook-Fundamentals*. 1997. Atlanta: American Society of Heating, Refrigerating, and Air-Conditioning Engineers, pp. 19.75-19.79.

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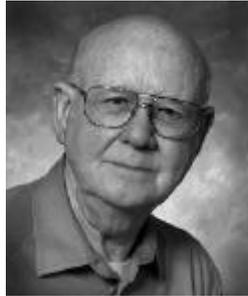
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