

# Nanoengineering — Toward Controlled Creation of New Inorganic and Composite Material Nanostructures

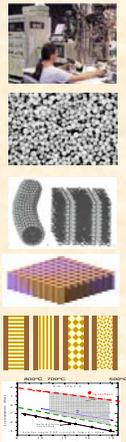
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## Outline

- **General introduction**
  - Our interests and facilities in nanomaterials
  - Nanoengineering
- **Chemical engineering of nanoparticles and material nanostructures**
  - Liquid phase synthesis
  - Processing by external fields
- **“Molecular engineering” of nanostructures**
  - Arrays of inorganic nanochannels/nanotubes: orientation/ordering control
  - Space-confined self-assembly: nanostructure control within inorganic nanowires
- **Potential applications**
  - Fuel cells, solar cells, sensors, ...
- **Concluding remarks**

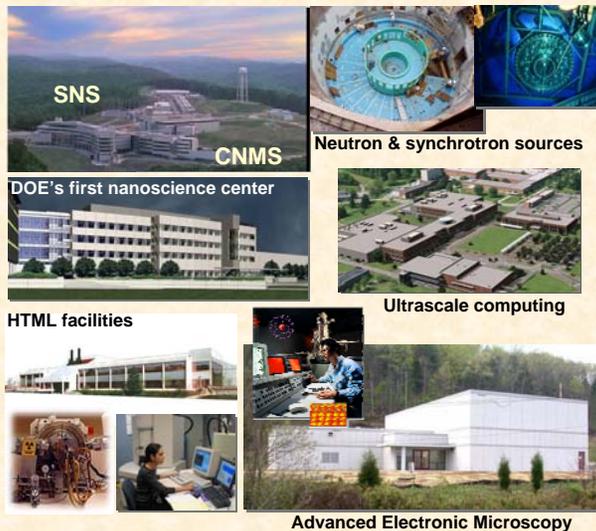


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## ORNL Is A World Leader in Nanosciences and Nation’s Premier Facilities in Advanced Materials Research

- **Center for Nanophase Materials Sciences (CNMS)**
- **Spallation Neutron Source (SNS) & High Flux Isotope Reactor (HFIR) scattering facilities**
- **High-Temperature Materials Laboratory Facilities (HTML)**
- **(NEW!) Center for Radiation Detection Materials and Systems (CRDMS)**



## Oak Ridge National Laboratory



**Excellence in Science and Innovative  
Solutions to Complex Problems**

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# Engineering vs. Science

• Definition (per Webster's):

- **Science:** "a branch of knowledge or study dealing with a body of facts or truths systematically arranged and showing the operation of general laws ... systematic knowledge of the physical or material world ..."
- **Engineering:** the art or science of making practical application of the knowledge of pure sciences, as physics, chemistry, biology, etc. ...
- **Nanoengineering:** ???
  - Must be enabling, not just the scale-up problem
  - Beyond existing chemical engineering, materials engineering, etc.
  - Need a transition from nanoscale science to nanoengineering



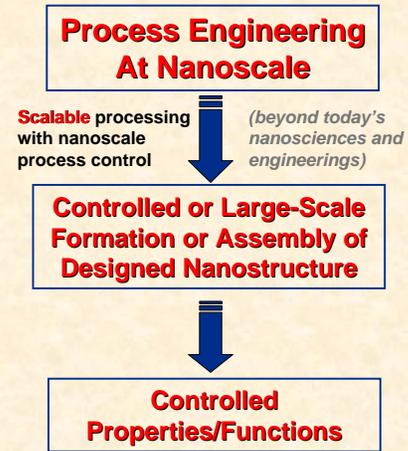
**NanoTechnology shall go beyond NanoScience !**

- Process scale
- Level of precision of process control
- Production through-put rate
- Dimension of products

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# Our Interests: Develop "NanoEngineering" for Controlled Nanosize & Nanostructure

- **Chemical processes** (including biomimetic processes)
  - "bottom-up" in nature
  - suitable for large-scale production economically
- **Engineered Chemical Processing** of nanomaterials - a new paradigm
  - Nanoscale chemical reaction and process engineering (chemical engineering)
  - Molecular engineering (chemistry + self-assembly)
  - Interface engineering



# Molecular Design and Process Engineering Control the Growth of Nanostructure

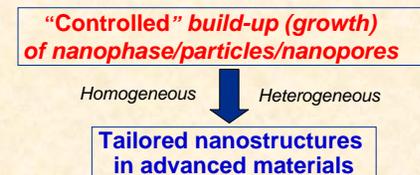
examples of fundamental building blocks

<p><b>Particle</b> Size &amp; distribution Shape Crystallinity Surface chemistry Pore structure Core-shell Colloidal crystal Linked network</p>	<p><b>Film</b> uniformity Located growth on surface Nanophase grain size Surface roughness Crystal structure Patterned film Film porosity Multilayer</p>	<p><b>Porous inorganics</b> Nanopore size Shape <b>Orientation</b> Domain scale Nanorod,wires Nanocomposites</p>

**Chemical process understanding and control must be obtained to achieve nanoscale engineered build-up of materials**

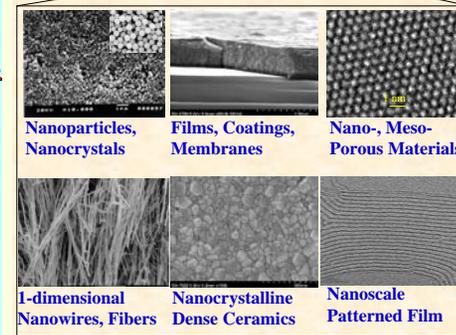
# Nucleation and Growth Is An Enabling Approach to Build Up Advanced Nanomaterials

- Allow *bottom-up* manufacturing via nucleation and growth
- A world of inorganic materials can be made by controlled "living growth"



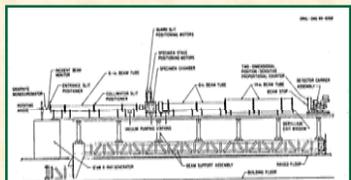
**Engineered manufacturing of advanced ceramics/composites could impact various applications**

- Energy storage and conversion (fuel cells, solar cells)
- Separations (inorganic membranes)
- Catalysis
- Micro-/nano-electronics
- Sensors, devices
- Targeted therapy, drug delivery, etc.



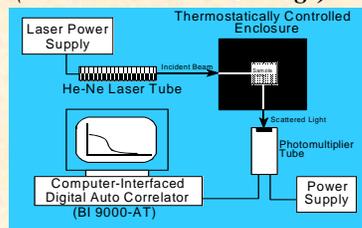
We Focus on Real-Time Process Monitoring and Control to understand & develop engineering processes for manufacturing nanomaterials

- Small-Angle Scattering (SAXS and SANS) (0.2 – 200 nm length scale for SAXS)



$$I(Q) = I_0(Q) N_p (\Delta\rho)^2 \exp(-Q^2 R_g^2/3)$$

- Dynamic Light Scattering (DLS) (5 – 1000 nm nanosize range)



- In-situ, high-temperature XRD (HTXRD)



HTML facility offer existing capability

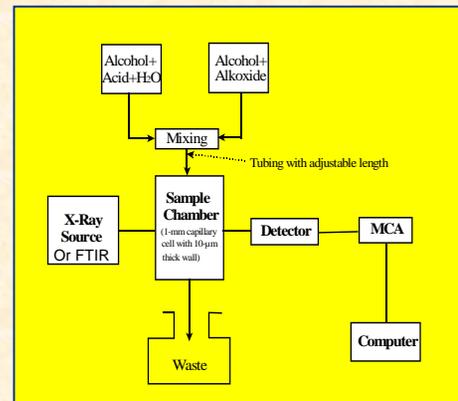
Scattering is complementary to electron microscopy  
**HRTEM**



Example: SAXS Monitoring of Sol-Gel Processing of Nanomaterials in Solutions

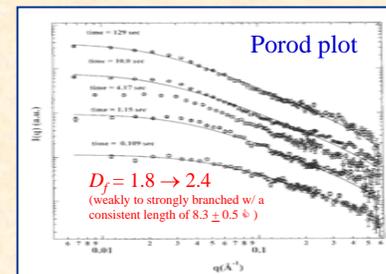
We have developed capability for studying early stages (as short as 80 ms).

SAXS from Early-Stage Zr-Nanoclusters



A "Steady-State" Technique Based on Rapid Mixing Flow-Cell

*J. Mater. Sci.*, 35, 1957 (2000).  
*J. Non-Crystal. Solid*, 246, 197 (1999).



Mass fractal scattering model

$$I(q) = NP(q) S(q)$$

$N$  = number density  
Form factor (for spherical particles):  
 $P(q) = V^2 (\Delta\rho)^2 \{3[\sin(qr_p) - qr_p \cos(qr_p)] / (qr_p)^3\}^2$   
Structure factor:  $S(q) = [1 + 1/(qr_p)^D D_p \Gamma(D_p - 1) \sin(D_p - 1) \arctan(q\xi) / 1 + 1/(q^2 \xi^2)]^{D_p - 1/2}$

Outline

- General introduction
- **Chemical engineering of nanoparticles and material nanostructures**
- Molecular "engineering" of nanostructures
  - Array of nanochannels/nanotubes: orientation/ordering control
  - Space-confined self-assembly: nanostructure control within nanowires
- Potential Applications
  - Fuel cells, solar cells, sensors, ...
- Concluding Remarks

Example: Engineered Processes Need to be Developed for Novel DTS Processing of Monodispersed Oxides

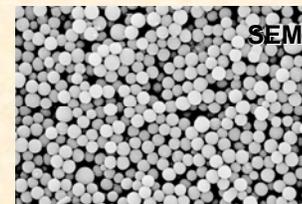
Highlight of Dielectric-Tuning Synthesis (DTS)



*J. Colloid Inter. Sci.*, 222, 20 (2000).  
*J. Amer. Ceram. Soc.*, 82, 2313 (1999).

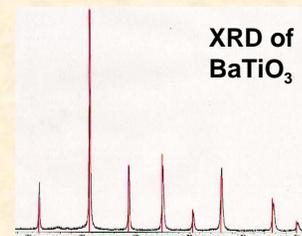
Oxides: e.g. ZrO<sub>2</sub>, TiO<sub>2</sub>, BaTiO<sub>3</sub>, and ZrTiO<sub>4</sub>  
**Uniqueness:**

- use inorganic precursors
- low temperature synthesis
- different from sol-gel method
- Capable of sol-gel processing



**Features:**

- Size: near-monodispersed
- Phase: nanocrystalline or amorphous
- High purity
- Nanoscale homogeneity
- Controllable morphology



*J. Mater. Sci.* 35, 1 (2000).  
*Powder Technol.*, 110, 2 (2000).

## Scale-up of DTS chemical process for nanoparticle synthesis

- Investigate engineering science issues: scale up, transition to continuous production

Glass vials (10-50mL)



Engineering reactors (5000mL, better control in T, P, mixing parameters)

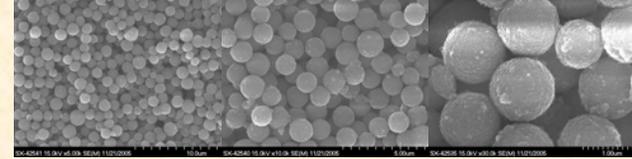


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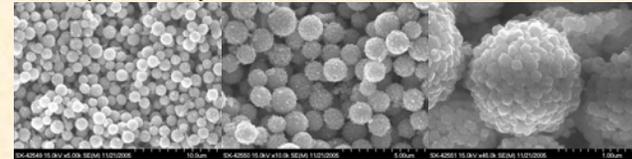
UT-BATTELLE

Technology for engineering scale production of nano-particles are still being developed for future energy generation and storage

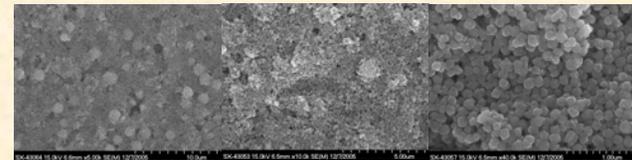
(A) Nanocrystalline TiO<sub>2</sub> Microspheres



(B) Nanocrystalline Microsphere Assemblies of BaTiO<sub>3</sub>



(C) Dispersed Nanocrystals of BaTiO<sub>3</sub> by Designed Dis-aggregation from Microsphere Assemblies



## New Arenas: Engineering Control of the Size and Nanostructures

Current nanoscience effort is restrained inside this box

Chemistry, materials science and engineering, etc.

“Outside-the-Box”:

“engineered” monitoring and control of processes

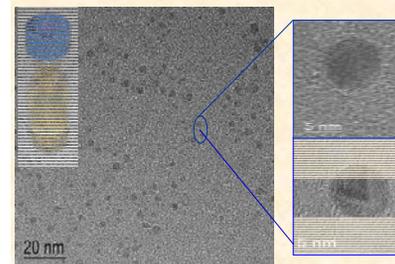
- Laser-assisted processing
- Field-directed processing (Electrical, Magnetic, Microwave...)
- Electrochemical processing
- Thermal processing
- Molecular assembly-assisted processing

*Diversified physical/chemical means of engineering controls enable*

- novel ways of manufacturing
- creation of new nanostructures
- better control of process parameters

## Laser Processing: Induced Nanocrystal Homogenization

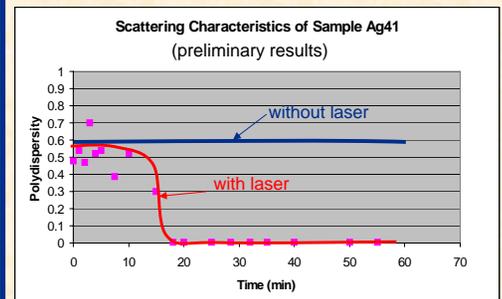
“Naked” Metallic Nanocrystals by A Thermal Electrochemical Process



This process, developed at ORNL, generates possibly a new class of Ag nanocrystals.

- Size < 10 nm
- Free from any organic capping molecules
- Colloidally stable

**2 orders-of-magnitude reduction in polydispersity! (from 0.5 to 0.005) while size does not change**



- Polydispersity: a parameter in dynamic light scattering measurement to quantify particle size distribution.
- < 0.02, for monodisperse or nearly monodisperse samples
  - 0.02-0.08, for narrow size distributions
  - > 0.08, for broader distribution

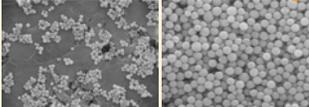
# Field-Coupled Chemical Processing

(Engineered Tailoring of Nanocluster Growth/Assembly Needs More R&D)

## Programmable Microwave Hydrothermal Reactions

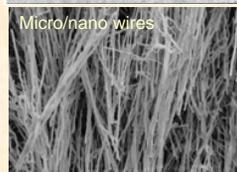
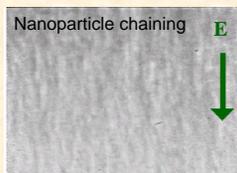


Faster heating      Slower heating



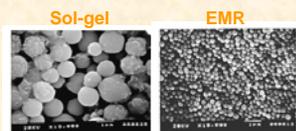
- Microwave volumetric heating **overcomes heat transfer limitations**
- Particle size can be controlled by manipulating the nucleation rate
- Achieving computer programmed synthesis

## Growth of Microfibrils in Electric Field



- field induce polarization and **oriented growth**
- Inorganic long "hair":
  - morphology/diameter can be tailored

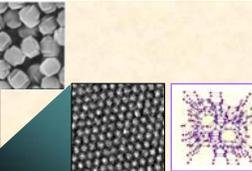
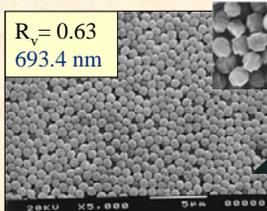
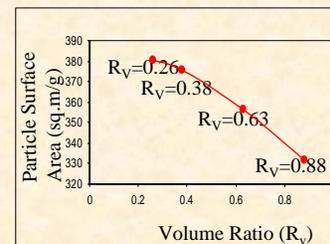
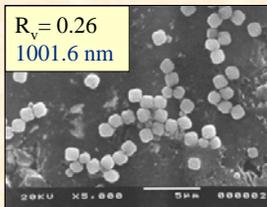
## Electrohydrodynamic Mixing Reactor (EMR)



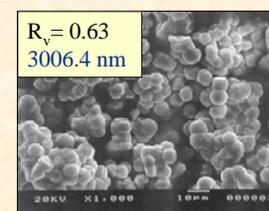
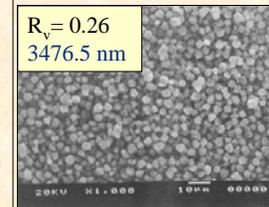
- Improved nanoscale mixing **overcomes mass transfer limitations**; allows
  - greater control of nucleation
  - synthesis beyond sol-gel processing limit

# Microwave Processing Offers More Flexibility than Conventional Hydrothermal Processing

Synthesis Temp = 150°C



Synthesis Temp = 180°C



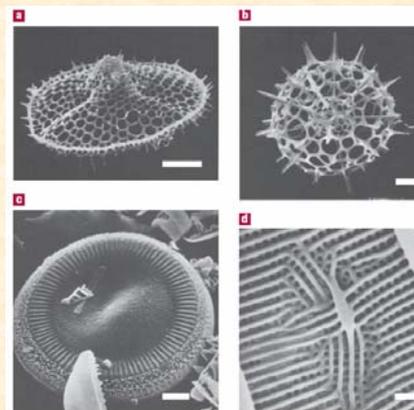
Various changes in zeolite particle size and morphology can be realized.

## Outline

- General introduction
  - ORNL nanoscience, research facilities, etc.
  - Our research interests and expertise
- Liquid-phase synthesis of nanoparticles
- Processing engineering of material nanostructures
- **Molecular "engineering" of inorganic nanostructures – a new paradigm**
  - Array of nanochannels/nanotubes: orientation/ordering control
  - Space-confined self-assembly: nanostructure control within nanowires
- Potential Applications
  - Fuel cells, solar cells, sensors, ...
- Concluding Remarks

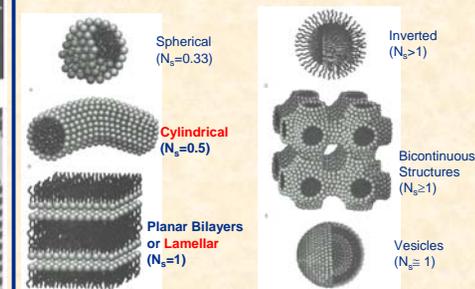
## Natural Beauty and Diversity of Self-Assembled Nanomaterials

Silicic skeletons of unicellular organisms:  
a, b – Radiolaria; c, d - diatoms



Sanchez et al., Nature Mater. 2005

**Molecular self-assembly** is the spontaneous and reversible organization of building-blocks (molecules) under thermodynamic equilibrium conditions into structurally well-defined and rather stable arrangements through a number of **noncovalent interactions**. (Shuguang Zhang 2002, Tirrell & Katz 2005)

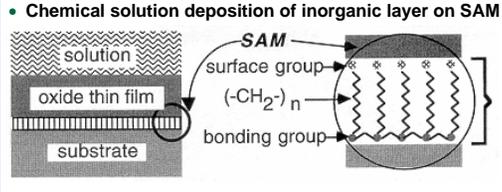


Hiemenz, Principles of Colloid and Interfacial Sciences

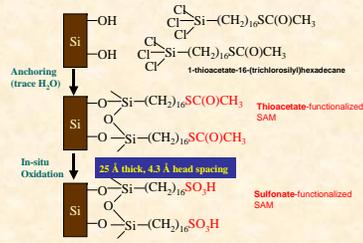
• **Biomimetism/bioinspiration** → Design/creation of innovative materials/systems

# Molecular Design-Based Surface Deposition and Interface Film Growth of Oxides

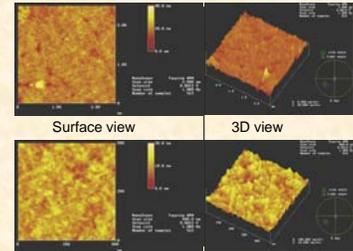
- Modify substrate with organic self-assembled monolayers (SAM)
- Chemical solution deposition of inorganic layer on SAM



- Important for film growth and interface engineering in nanocomposites
- We have grown and studied various oxide deposition (ZrO<sub>2</sub>, HfO<sub>2</sub>, etc.)
- The approach has induced homogeneous deposition and improved
  - uniformity, roughness
  - coherence, adherence
  - possible selective deposition

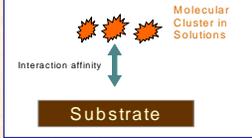


- Inter. J. Appl. Ceram. Technol., April (2004)  
 - Solid State Ionics, 151, 69 (2002).

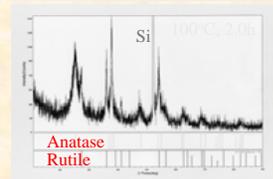
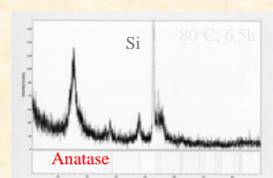
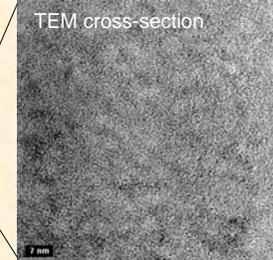
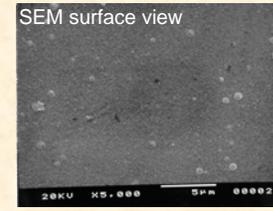
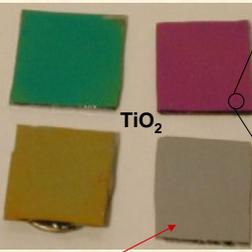


# Chemical Processing May Control Thickness and Nanostructure of TiO<sub>2</sub> Film

**Strategy:**  
 Design of Molecular Complex Ti Species or Nanoclusters in Bulk Solutions



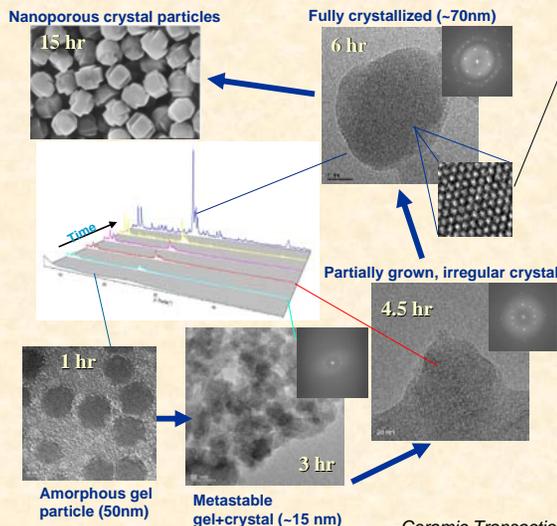
(Colorful mirror-like films)



- Nanocrystalline TiO<sub>2</sub> films (containing ~3 nm nanocrystals)
  - high refractive index (2.2)
  - large band-gap (up to 3.7 eV)
  - low surface roughness

Application to Solar Cells?

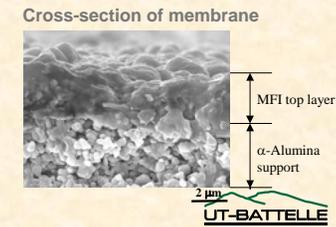
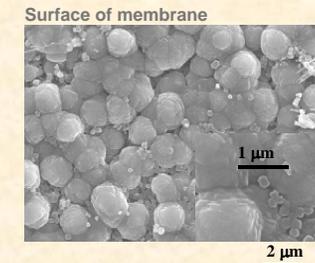
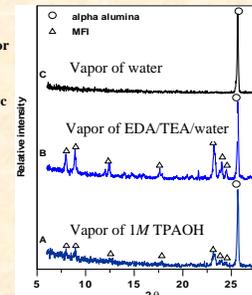
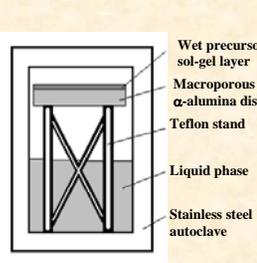
# Molecular-Templated Growth and Crystallization of Zeolite Particles



- A good model system for understanding large-scale process but with possible achievement of nanoscale homogeneity
  - Sub-nm micropores are created by molecular templating inside a crystal particle
  - Small template molecules control the size, shape, and uniformity of the pores
- Still needs further development for efficiency, selectivity, and cost saving for practical applications in catalysts and adsorbents

Ceramic Transactions, 137, 3-21 (2003).

# A Novel Vapor-Phase Thermal Conversion Synthesis of MFI Zeolite Membranes



Results of zeolite membrane synthesis under different conditions.

Synthesis sol-gel on substrate	Liquid phase	Top layer	Thickness (μm)	Quality
7.0% SiO <sub>2</sub> + 2.1% TPAOH + 0.8% NaOH + 90.1% H <sub>2</sub> O	1 M TPAOH	MFI	3	Good
7.0% SiO <sub>2</sub> + 2.1% TPAOH + 0.8% NaOH + 90.1% H <sub>2</sub> O	EDA/TEA/H <sub>2</sub> O (8.1/30.7/61.2)	MFI	12	Poor
7.0% SiO <sub>2</sub> + 2.1% TPAOH + 0.8% NaOH + 90.1% H <sub>2</sub> O	H <sub>2</sub> O	Amorphous	---	---
7.0% SiO <sub>2</sub> + 2.9% NaOH + 90.1% H <sub>2</sub> O	1 M TPAOH	NaP-cubic	---	---
7.0% SiO <sub>2</sub> + 2.9% NaOH + 90.1% H <sub>2</sub> O	EDA/TEA/H <sub>2</sub> O (8.1/30.7/61.2)	ANA-cubic	---	---

J. Mater. Sci., 38 (5): 979 (2003).

# Rich Morphologies via Self Assembly and Organization

Park C. et al., Polymer (2003)

Park C. et al., Polymer (2003)

Nature of patterns	Spheres (SPH) (3D)	Cylinders (CYL) (2D)	Double gyroid (DG) (3D)	Double diamond (DD) (3D)	Lamellae (LAM) (1D)
Space group	$Im\bar{3}m$	$p6mm$	$Ia\bar{3}d$	$Pn\bar{3}m$	$pm$
Blue domains: A block					
Volume fraction of A block	0-21%	21-33%	33-37%	33-37%	37-50%

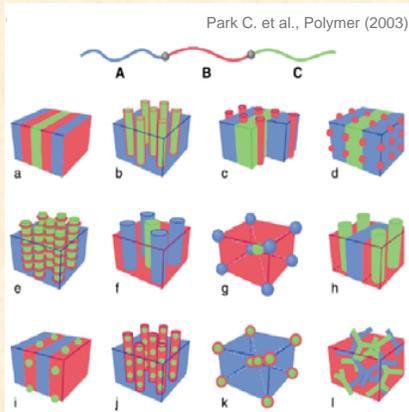
Schematic phase diagram showing the various "classical" BCP morphologies adopted by non-crystalline linear diblock copolymer. The blue component represents the minority phase and the matrix, majority phase surrounds it.

## Polymorph selection in porous block copolymers

Ward, Hillmyer, J. Am. Chem. Soc., 126, 3382 (2004)

## Tailored pore wall functionality

Hillmyer et al., Macromolecules, 38, 3 (2005)  
ibid., Faraday Disc. 128, 149 (2005)



Schematic of morphologies for linear ABC triblock copolymer. A combination of block sequence (ABC, ACB, BAC), composition and block molecular weights provides an enormous parameter space for the creation of new morphologies. Microdomains are colored as shown by the copolymer strand at the top, with monomer types A, B, and C confined to regions colored blue, red, and green, respectively.

# Molecular Self-Assembly Processing of Inorganic Nanomaterials -- Grand Challenges

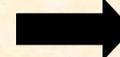
(a) Current state of the art in molecular templated synthesis

(b) Future realization of large-scale ordering and control of orientation and wall function

Powders containing randomly oriented small domains of ordered mesopores



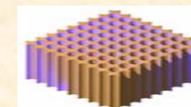
Beyond what nature can create



Film Layers containing randomly oriented, disordered mesopores



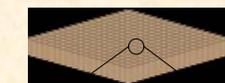
Oriented Nanopores (1-100 nm dia.)



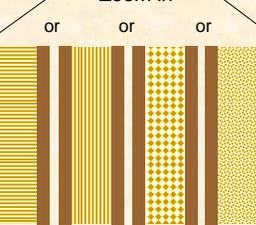
Parallel pore channels is in perpendicular to the layer or substrate surface



Guest-Host Nanocomposite Superstructures



Zoom in



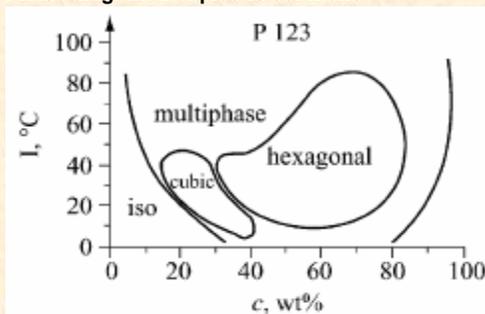
Nanostructure control inside nanowire

Enhanced or New Functions

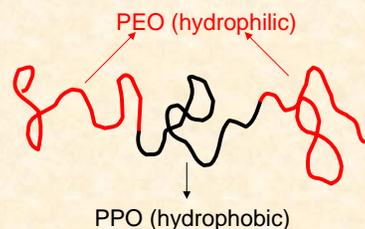
# Self-Assembly of P123

P123:  $(PEO)_{20}(PPO)_{70}(PEO)_{20}$

Phase diagram in aqueous solutions



Wanka G. et al., Macromolecules 27, 4145 (1994).

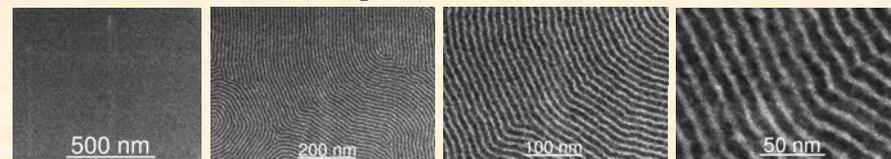


Hydrophobic effect in dilute aqueous solution causes the self-assembly

Molecular Design-Based Build-up

# Developing Hybrid Molecular Self-Assembly of Oxide Mesostructures

Formation of HORIZONTAL  $SiO_2$  MESOSTRUCTURE via block copolymer



Z-contrast STEM images: Bright region representing silica. Channels (7.5-nm gap) exhibit a highly connected horizontal structure. (V. F. de Almeida, D. A. Blom, L. F. Allard, M. Z. Hu, S. Dai, C. Tsouris, Z. Zhang, Microscopy and Microanalysis 2003, San Antonio, Texas, August 3-7, 2003.)

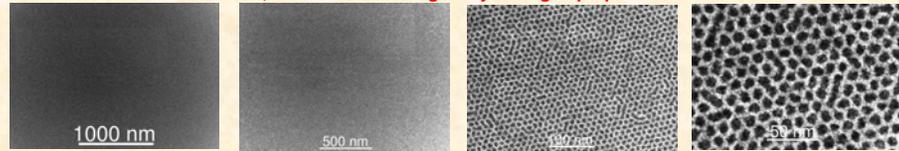
## Key Issues:

- Orientation control: let meso-channels standing up normal to substrate?
- Expanding domain size of ordering
- Structure stability

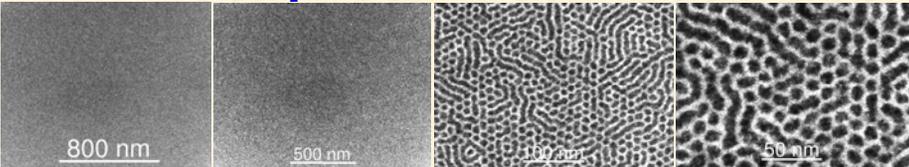
Molecular Engineering-Based Build-up

# Process Engineering Enables Channel Orientation in Nano/Mesoporous Films

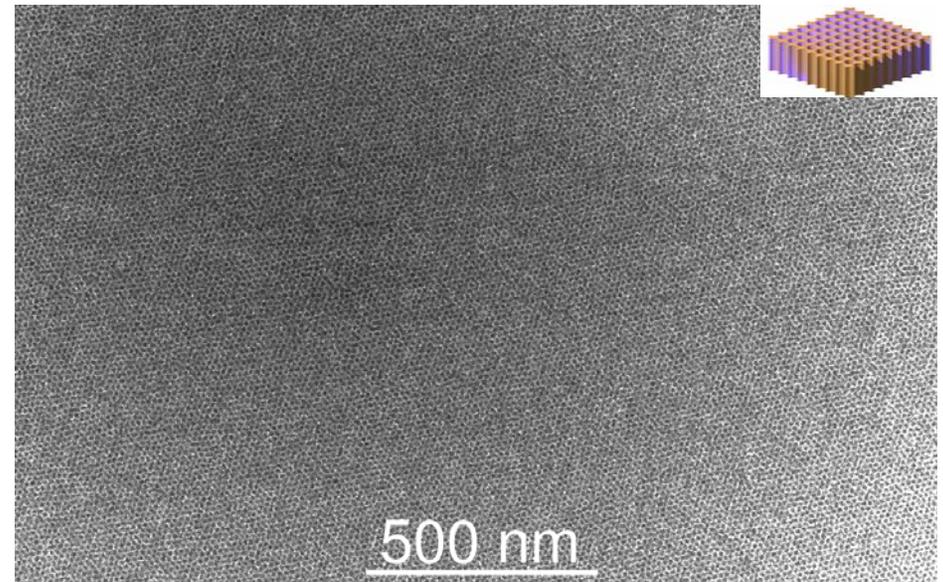
•Achieved Vertical SiO<sub>2</sub> Mesostructure (black dots are channel pores, bright region-SiO<sub>2</sub>)  
Pore size diameter of 7.5 nm; channels are hexagonally arranged perpendicular to the film surface.



•Observed Transition SiO<sub>2</sub> Mesostructure (from vertical to some degree of horizontal connectivity)



But, the engineering sciences (that control the mechanisms of channel orientation, control of aspect ratio and pore connectivity, ordering scale-up, etc.) still need to be developed !



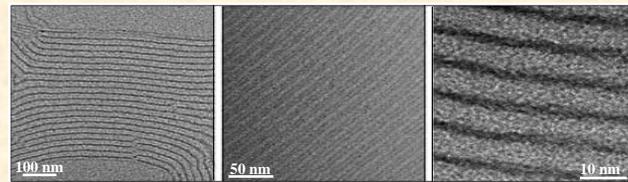
- Very large area of hexagonal ordering: achieved mm range
- Possible hard template for arrays of nanoscale vertical channel columns
- Many potential applications (fuel cells, solar cells, sensors, catalysts, etc. )

Molecular Design-Based Build-up

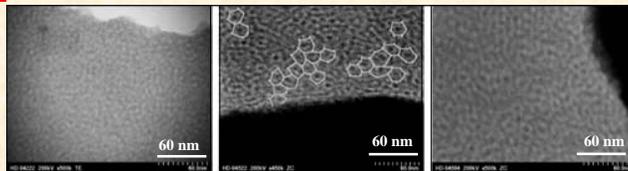
# Understanding Controlled Self-Assembly: Mesopore Orientation in Many Other Oxides

High-resolution TEM images of titania layer mesostructures containing parallel pore channels with spacing of ~10nm. Dark pore channel lines correspond to titania phase.

TiO<sub>2</sub>



STEM images of mesostructured titania showing distorted hexagonal phases (TE-transmission mode; ZC-Z-contrast mode, brighter areas correspond to titania phase).



(Hu, unpublished results)



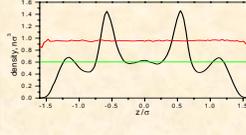
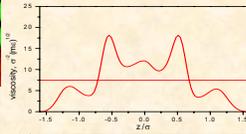
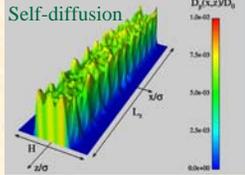
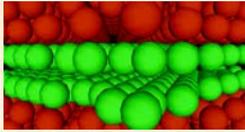
Nano-in-Nano Control: Space-Confining Self-Assembly of Organic-Inorganic Wires in Nanopore Channels (0.5 – 200nm)

# Self Assembly in Confined Pore Channels: Guest-Host Superstructures

## Potential Applications

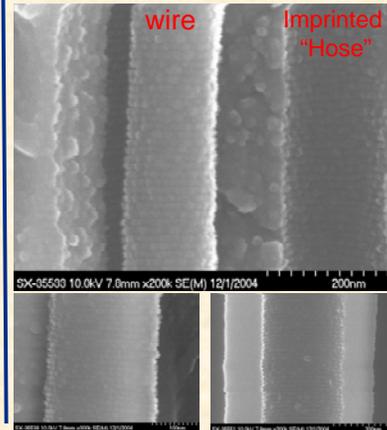
Electronic materials, Fuel-cell electrolyte, Optical materials, Magnetism, Catalysts, Nanocomposites

## Results from Statistical Mechanical Theory and EMD simulation

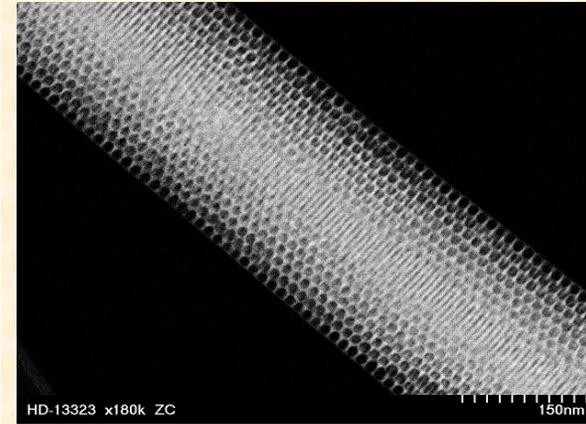


*J. Nanosci. Nanotechnol.* **2**, 209 (2002).  
*Ceram. Trans.* **137**, 101 (2002).

- Wires inside pore channels consist of oriented, parallel stacked layers of oxide (10-nm thick)



# Revealing "layered" structure with STEM



"Layered" nanostructure (SEM)

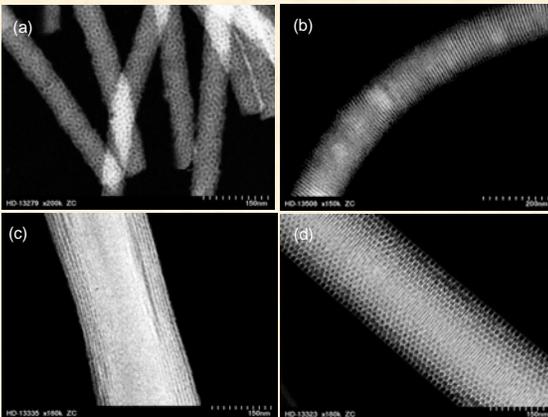


highly oriented and ordered hexagonal mesophases (STEM)

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# Molecular & Process Engineering Approach Could Enable Refined "Nano-in-Nano" Control of Nanostructures



Various oriented and ordered inorganic oxide nanostructures inside space-confined wires/channels:

- (a) Random oriented mesopores and small domain size of hexagonal phases,
- (b) Parallel lamellar phase oriented in perpendicular to the pore channel,
- (c) Lamellar phase oriented in parallel to the pore channel,
- (d) Highly ordered and oriented hexagonal mesopores (7.5nm dia.).

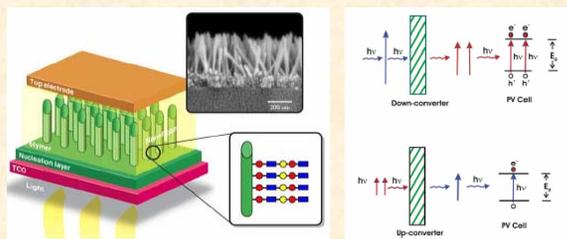
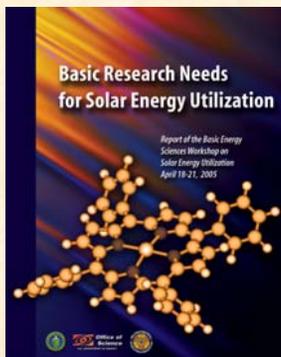
# Outline

- General introduction
  - ORNL nanoscience, research facilities, etc.
  - Our research interests and expertise
- Liquid-phase synthesis of nanoparticles
- Processing engineering of material nanostructures
- Molecular "engineering" of nanostructures
  - Array of nanochannels/nanotubes: orientation/ordering control
  - Space-confined self-assembly: nanostructure control within nanowires
- **Potential Applications**
  - Fuel cells, solar cells, sensors, ...
- Concluding Remarks

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## Success Could Directly Impact Energy Materials of the Future: Fuel Cells and Solar Cells



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## Concluding Remarks

- **Chemical engineering and “molecular engineering” of nanosize and nanostructure are frontiers for new discovery and creation of innovative materials.**
  - Chemical processing offers great opportunities in large-scale manufacturing and production while allowing control of material nanostructures and functions
- **At ORNL, we endeavor to develop new engineering sciences and manufacturing capabilities in nanomaterials for energy, security, and other applications**
  - Nanomaterials Engineering Laboratory: Focus on chemical processing and engineering
  - State-of-the-art instrument facilities on real-time process monitoring and nanoscale characterization (CNMS, SNS, ...)
- **Future nanofabrication demand “process understanding-based” control of**

**Size → Shape → Ordering → Orientation**

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## Acknowledgments

**Post-Docs:** L. Pozhar, J. Dong, A. Singhal

**Students:** P. Lai, X. Wang, L. Khatri, J. Zielke, S. Morton, A. DeBaille, V. Kurian, G. Miller, K. Booth, B. Grant, J. Clavier

**ORNL Colleagues:** C. Easterly, D. DePaoli, R. Hunt, A. Payzant, J.-S. Lin, C. Rawn, Allard, D. Blom, I. Anderson, G. Jellison, Y. Wei, P. Becher, T. Armstrong, C. Mattus, J. Wang, V. de Almeida

**Collaborators:** D.-L. Shi, M. T. Harris, D. Green, B. Lee, Y.-S. Lin, M. DeGuire, N. Xu

- Research sponsored by
  - Division of Materials Sciences, Material Chemistry Program, Office of Sciences, U.S. Department of Energy (DOE)

- ORNL HTML facility

*Oak Ridge National Laboratory is managed by UT-Battelle, LLC, for the U.S. Dept. of Energy under contract DE-AC05-00OR22725*

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## Overview: Process Engineering and Molecular Engineering Enable Tailored Chemical Processing of Diversified Nanostructures

